

Synthesis of Mirasorvone, a Defensive Steroid from the Sunburst Diving Beetle (*Thermonectus marmoratus*)

Zhi-Cai Yang and Jerrold Meinwald*

Department of Chemistry, Baker Laboratory, Cornell University, Ithaca, NY 14853

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Abstract: The structure and stereochemistry of mirasorvone, an 18-oxygenated pregnane from the defensive prothoracic glands of the sunburst diving beetle, Thermonectus marmoratus, has been confirmed by synthesis and single-crystal X-ray diffraction analyses. © 1998 Elsevier Science Ltd. All rights reserved.

Mirasorvone, an insect defensive steroid, is the chief component of the milky fluid produced by the defensive prothoracic glands of the sunburst diving beetle, *Thermonectus marmoratus*. This steroid was characterized as the hemiketal 1 based on an analysis of extensive spectroscopic data. To the best of our knowledge, this is the first 18-oxygenated steroid to be isolated from an insect source. In order to confirm the structure and stereochemistry of mirasorvone, as well as to make available sufficient material for a study of its behavioral and physiological activities, we have carried out the synthesis of 1 described in this report.

The synthetic route to mirasorvone (1) is outlined in Scheme 1. 3β -Acetoxypregn-5-ene- 20β -ol (2) was prepared by reduction of the commercially available pregnenolone acetate with sodium borohydride in THF and methanol (1:1) in 85% yield.² Irradiation of 2 in the presence of iodine, freshly recrystallized lead tetraacetate,

Aco
$$\frac{1}{2}$$

Aco $\frac{1}{3}$

Aco

Scheme 1

1

7

Reagents and conditions: i. Pb(OAc)₄, I₂, CaCO₃, cyclohexane, *hv*, reflux; ii. CrO₃, H⁻, acetone, 0°C, 10 min., 50% from 2; iii. AgOAc, 1,4-dioxan, H₂O (9:1), 65°C, 12h, 92%; iv. 10% NaOH, MeOH, r.t., 30 min., 95%; v. Al(OPr')₃, N-methyl-4-piperidone, PhMe, reflux, 5h, 72%; vi. chloranil, *tert*-butanol, reflux, 30 min., 54%.

and dried calcium carbonate in cyclohexane, using a 500 w-tungsten lamp,^{3,4} was carried out until the iodine color had disappeared (ca. 25 min.). The reaction mixture was immediately quenched and worked up to afford a residue of 3 β -acetoxy-18-iodopregn-5-ene-20 β -ol, which was oxidized (without purification) with Jones reagent to give the unstable 3 β -acetoxy-18-iodopregn-5-ene-20-one (3) in 50% yield. Reaction of the iodoketone 3 with silver acetate in aqueous dioxane³ provided the 18, 20-hemiketal 4 in 92% yield, m.p. 157-158°C, [α]_D = -16° (c, 0.2, CHCl₃)[lit.⁵ m.p. 158-161°C], while silver-ion-promoted solvolysis in methanol

gave the corresponding 18,20-epoxy- 20α -methoxy derivative 5, m.p. $174-176^{\circ}$ C, $[\alpha]_D = +22.6^{\circ}$ (c, 0.3, CHCl₃)[lit.⁴ m.p. $171.5-175^{\circ}$ C]. The configuration of 5 at C-20 was established by X-ray diffraction analysis⁶ (Fig. 1). Another X-ray diffraction analysis⁶ showed that the 18,20-hemiketal 4 also crystallizes in the 18,20-epoxy- 20α -hydroxy form (Fig. 2).

Fig. 1: X-ray crystallographic structure of 5

Alkaline hydrolysis of the 3β-acetate 4 afforded the 3β-alcohol 6 in 95% yield, m.p. 116-118°C, $[\alpha]_D = +50^{\circ}(c, 1.3, CHCl_3)$. A Keana-Reich modification of the Oppenauer oxidation^{4,7} of the alcohol 6 with aluminum isopropoxide and N-methyl-4-piperidone in dry toluene provided the enone 7 in 72% yield, m.p. 160-161°C, $[\alpha]_D = +158^{\circ}$ (c, 0.3, CHCl₃){lit.⁴ m.p. 154-156°C; lit.⁸ m.p. 164-165°C, $[\alpha]_D = +157^{\circ}$; lit.⁹ m.p. 159-160°C, $[\alpha]_D = +159^{\circ}$ }. Finally, dehydrogenation of the enone 7 with chloranil in *tert*-butanol¹⁰ gave the target mirasorvone (1) in 54% yield, m.p. 138-142°C, $[\alpha]_D = +180^{\circ}$ (c, 0.1, CHCl₃). The 20α-hydroxy configuration of the hemiketal 1, assigned on the basis of NMR spectroscopic data, seems to be generally preferred by these 18-hydroxy-20-ketopregnane derivatives, since the analogous structures 4, 5, and 7^6 also possess this stereochemistry (Fig. 1-3).

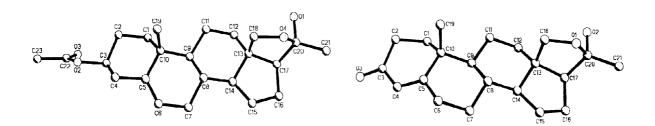


Fig. 2: X-ray crystallographic structure of 4

Fig. 3: X-ray crystallographic structure of 7

The only steroidal hemiketal closely related to mirasorvone whose C-20 configuration has been definitively determined appears to be aldosterone (8), "the most potent regulator of electrolyte excretion and ... vital ... to innumerable life processes." Interestingly, aldosterone has a configuration at C-20 opposite to that of mirasorvone. 11,12

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References and Notes:

- Meinwald, J.; Huang, Q.; Vrkoč, J.; Herath, K. B.; Yang, Z. C.; Schröder, F.; Attygalle, A. B.; Iyengas, V. K.; Morgan, R.; Eisner, T. Proc. Natl. Acad. Sci. USA (in press).
- 2. Heusler, K.; Wieland, P.; Meystre, Ch. Org. Syn. Coll. Vol. V, 692-698. John Wiley and Sons, Inc., New York, 1973.
- 3. Kirk, D. N.; Rajagopalan, M. S. J. Chem. Soc. Perkin Trans. I 1975,1860-1864.
- 4. Kirk, D. N.; Rajagopalan, M. S. Steroids 1976, 27, 269-274.
- Meystre, C.; Wettstein, A.; Jeger, O.; Anner, G.; Heusler, K.; Wieland, P. Swiss Pat., 410,936/1966 (Chem. Abstr. 1967, 66, 65,745d).
- 6. The crystal of 4 was in the orthorhombic system with space group $P2_12_12_1$ and the unit cell dimensions were determined as a = 6.4045(13), b = 9.958(2), c = 32.484(7) Å, $\alpha = \beta = \gamma = 90^{\circ}$, V = 2071.9(7) Å³, Z = 4; 5 was in the orthorhombic system with space group $P2_12_12_1$ and the unit cell dimensions were determined as a = 6.6010(13), b = 12.444(3), c = 26.377(5) Å, $\alpha = \beta = \gamma = 90^{\circ}$, V = 2166.7(7) Å³, Z = 4; 7 was in the orthorhombic system with space group $P2_12_12_1$ and the unit cell dimensions were determined as a = 6.3360(13), b = 11.419(2), c = 25.230(5) Å, $\alpha = \beta = \gamma = 90^{\circ}$, V = 1825.4(6) Å³, Z = 4. All atomic coordinates, bond lengths and angles and thermal parameters have been deposited at the Cambridge Crystallographic Data Center.
- 7. Keana, J. K. W.; Reich, R. Synth. Comm. 1972, 2, 323-325.
- 8. Wehrli, H.; Cereghetti, M.; Schaffner, K.; Jeger, O. Helv. Chim. Acta 1960, 43, 367-371.
- 9. Buzzetti, F.; Wicki, W.; Kalvoda, J.; Jeger. O. Helv. Chim. Acta 1959, 42, 388-390.
- 10. Agnello, E. J.; Laubach, G. D. J. Am. Chem. Soc. 1960, 82, 4293-4299.
- 11. Duax, W. L.; Hauptman, H. J. Am. Chem. Soc. 1972, 94, 5467-5471.
- 12. This is paper no. 154 in our series, *Defense Mechanisms of Arthropods*; paper no. 153 is F. Schröder *et al.*, *Science* (submitted).